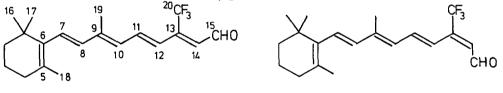
PREPARATION OF 20,20,20-TRIFLUORORETINAL (13-TRIFLUOROMETHYLRETINAL)

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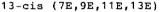
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Summary: All-trans-20,20,20-trifluororetinal (1) was prepared by two ways.

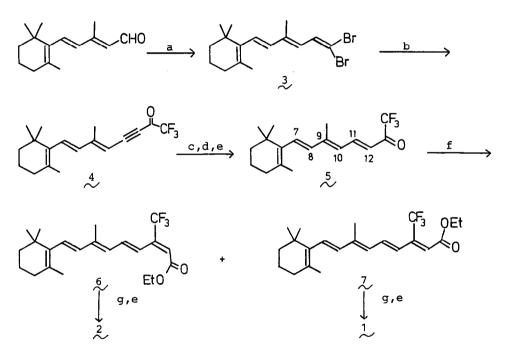
In a previous study on the absorption spectra of visual pigment and bacteriorhodopsin, the external point-charge model was proposed to accout for "opsin shift".¹⁾ In order to check the validity of this model, the analogs of retinal were prepared.²⁾ In this context, it is considered to be significant to prepare 20,20,20-trifluororetinal (1), since the fluorine substituent, characterized by strong electronegativity and small atomic size, may alter the electronic but steric properties of retinal. Although Gärtner et al.³⁾ have reported the synthesis of 1 (all-trans=7E,9E,11E,13Z), the structure he proposed for this compound is at variance with that which Liu et al.⁴⁾ determined on the basis of their experimental results. That is, they obtained the 13-cis isomer (7E,9E,11E,13E) (2) under essentially the same conditions.



all-trans (7E,9E,11E,13Z)

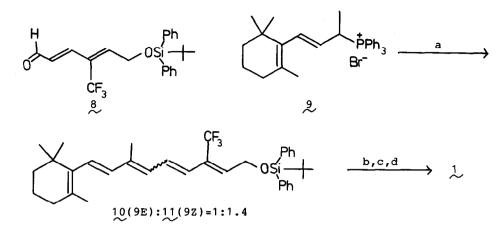


In the present paper, we report the synthesis of 1 and 2. An attempt was made to synthesize through the versatile intermediate (4) by which the preparation of 11-cis stereoisomer of 1 should be possible. Treatment of β -ionylideneacetaldehyde with CBr_4 -Ph₃P gave chain-extended dibromoolefine (3)⁵ in more than a 90% yield, which was converted to acetylenic ketone (4)⁶ in a 64% yield by the reaction of n-BuLi (2 eq.) and CF_3COOEt . The triple bond was reduced with Red-A1 [sodium bis(2-methoxyethoxy)aluminum hydride] after ketone (4) was reduced with NaBH₄. The alcohol function was oxidized back to ketone with MnO₂ to give compound (5) (51% yield from 4). The ¹H-nmr spectrum of 5 confirmed the all-trans structure: δ 6.20 (d, J=16.5Hz, H₈), 6.27 (d, J=12Hz, H_{10}), 6.42 (d, J=15Hz, H_{12}), 6.52 (d, J=16.5Hz, H_7), 8.04 (dd, J=15, 12Hz, H_{11}). To extend 5 by two additional carbons, we carried out the Peterson olefination of this compound. The reaction proceeded smoothly under mild conditions (-78 + -23°) (TMSCH₂COOEt/LICA/THF)⁷⁾ and the esters (6, 7) were obtained in a nearly 1:1 ratio (66%). These isomeric esters (6, 7) were easily separated from each other and assigned on the basis of the nmr spectrum.⁸⁾ Reduction by DIBAL and oxidation by MnO₂ of 6 and 7 gave 2 and 1,⁹⁾ respectively. The reported all-trans compound³⁾ was found not to be identical with our compound 1 but with compound 2 by an nmr spectra comparison.¹⁰⁾ No isomerization was expected to occur during the reduction and oxidation and thus, compound (1) was concluded to be all-trans-20,20,20-trifluororetinal (7E,9E,11E,13Z).



a: CBr₄-Ph₃P, b; 2eq. n-BuLi, CF₃COOEt in THF, c: NABH₄/EtOH d: Red-Al/Et₂O, e: MnO₂/CH₂Cl₂, f: TMSCH₂COOEt/LICA/THF, g: DIBAL/hexane

For confirmation of the structure of 1, its synthesis was carried out by a different route. During our study of trifluoromethylated diene syntheses, we prepared compound (8) by various methods.¹¹⁾ The Wittig reaction of 8 with phosphonium salt (9)¹²⁾ (nBuLi/Et₂O/-20°+rt) gave a mixture of stereoisomers 10 and 11 in a 51% yield (9E/9Z=1/1.4). Deprotection (nBu₄NF) of the silyl group generated a mixture of 9E- and 9Z-20,20,20-trifluororetinols (12,13), one of which, compound 12, showed behavior identical to that observed for the retinol obtained by the reduction of 7 (all-trans) in the HPLC experiments.



a: n-BuLi/Et₂O, b: n-Bu₄NF/THF, c: HPLC separation
d: MnO₂/CH₂Cl₂

The data from the above experiments clearly indicate that the 20,20,20-trifluororetinal reported by Gärtner et al.³⁾ is not actually all-trans compound (1) but 13-cis compound (2) (7E,9E,11E,13E).

References and Notes

Syntheses of all-trans-19,19,19-trifluororetinal and 20,20,20-trifluororetinal were carried out independently by Liu et al. at about the same time that our research was being conducted. D. Mead, A. E. Asato and R. S. H. Liu, 10P33, "'84 International Chemical Congress of Pacific Basin Societies ", Honolulu, Hawaii, 1984.

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- 2) Trifluoromethylated analogs of retinal and retinoic acid:
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- 4) Ref. 2c).
- 5) E. J. Corey and P. L. Fuchs, Tetrahedron Lett., 3769 (1972).
- 6) Ref. 5) and T. Kitazume, T. Sato and N. Ishikawa, p. 56, Abstracts of the ninth fluorine symposium in Hokkaido, Japan, 1983.

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- 8) The ¹H-nmr spectrum of 6 was quite similar to that of the methyl ester. Ref. 2b) and 2c). Compound 6 was eluted first in flash chromatography (silica gel, Hexane-EtOAc). ¹H-Nmr spectrum of 7: δ 6.12 (d, J=15.8Hz, H₈), 6.10 (d, J=12Hz, H₁₀), 6.12 (d, J=15.8Hz, H₁₂), 6.24 (S, H₁₄) 6.32 (d, J=15.8Hz, H₇), 7.07 (dd,

J=15.8, 12Hz, H₁₁).

- 9) The all-trans-20,20,20-trifluororetinal (1) was unstable and easily isomerized to 2 by HPLC (silica gel, Hexane-EtOAc) conditions.
- 10) The aldehyde proton of 1 appears at δ 10.07 (dq, J=7.7, 1.7Hz) in the dq coupling pattern due to H-H and H-F couplings. In the case of the aldehyde 2, the aldehyde proton at δ 10.14 (d, J=6.8Hz) showed no H-F coupling. This nmr coupling analysis is a very general way to determine the stereochemistry of the β -CF₃-substituted- α , β -unsaturated aldehyde system. ¹⁹F-Nmr spectrum of 1 (in ppm down field from external benzotrifluoride): 6.6 (bs).
- A part of this work has already been published: Y. Hanzawa, K. Kawagoe,
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- 13) The alcohol isolated by HPLC (Waters Liquid Chromatograph Model 590, μ -porasil, 5% dioxane in Hexane) was also oxidized with MnO₂ to give 1.

(Received in Japan 5 March 1985)